#### **Development of a Liquid Bridge Model for Particle Agglomeration and Defluidization in Plastic Pyrolysis**



*Subhodeep Banerjee*

*Research Scientist*





This project was funded by the United States Department of Energy, National Energy Technology Laboratory, in part, through a site support contract. Neither the United States Government nor any agency thereof, nor any of their employees, nor the support contractor, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.





#### **Subhodeep Banerjee1,2**

#### **<sup>1</sup>National Energy Technology Laboratory, 3610 Collins Ferry Road, Morgantown, WV 26505, USA**

#### **<sup>2</sup>NETL Support Contractor, 3610 Collins Ferry Road, Morgantown, WV 26505, USA**



### **Liquid Layer Development Proof-of-Concept**

- NATIONAL ERGY TECHNOLOGY **LABORATORY**
- Molten plastic that forms during the pyrolysis of plastic or municipal solid waste feedstock can lead to particle agglomeration A proof-of-concept of a liquid bridge model is developed based on Grohn et al.<sup>1</sup> and verification studies are completed<br>Rebound Capillary forces are • Capillary forces are based on droplet  $V_{\text{liq},i,z}$ Laplace pressure<sup>2</sup>  $V_{\rm drop}$ ideal wetting  $4\pi R\gamma \cos\theta$  $V_{\rm p,i,ini}$  $V_{\rm p,i,ini}$  $V_{\rm p,i,ini}$  $V_{\rm p,i,ini}$  $V_{\rm p,i,ini}$  $F_{cap,pp} =$ −1 $2V_b$  $V_{\text{liq,i,z}}$  $|u_{p,i,z+2}\>$  $1 + \binom{1}{2}$  $\frac{1}{2} - 1$  $V_{\text{liq},i,z+2}$ particle  $\mu_{p,i,z}$  $\pi Rh$ Time: 0.01 s  $V_{\text{b,}ij,z+1}$  $R \gg x$ , d, D  $u_{p,j,z+2}$  $u_{p,j,z}$  $r_2 \gg r_1$  $\phi$  small  $V_{\rm p,j,ini}$  $V_{\rm p,j,ini}$  $V_{\rm p,j,ini}$  $x = R \sin x$  $2r_1 \cos \theta$ Time:  $0.01$  s  $V_{\text{liq},\text{j},\text{z}}$  $V_{\text{liq},\text{j},z+2}$  $V_{b,ij,z+1} = V_{liq,i,z} + V_{liq,j,z}$  $V_{\text{b,}ij,z+1}/2 = V_{\text{liq},i,z+2} = V_{\text{liq},j,z+2}$ No liquid bridge 0.004  $\frac{1}{2}$ <br>  $\frac{1}{2}$ Time: 0.01 s Liquid bridge capillary force only ę. • Molten plastic droplet deleted on contact 0.001 with sand particle and mass and species transferred in perfectly inelastic collision  $\Omega$  $\overline{1}$  $\overline{2}$  $\overline{3}$  $\overline{4}$  $\overline{\mathbf{5}}$  $\Omega$ • Conservation of mass and momentum Time (s) L verified numerically

- No liquid bridge - Liquid bridge



#### **Spouted Bed Experiment/Model of Tang et al.**





**I.S. DEPARTMENT OF** 

ERGY



- Tang et al.<sup>3</sup> modeled their experiment using MFiX-DEM; simulations were run for 15 s and results averaged over final 10 s
- Drag model used is Beetstra<sup>4</sup>
- Capillary force

$$
F_{cp,n} = -\frac{2\pi\gamma R \cos\theta}{H/2d+1} - 2\pi\gamma R \sin\varphi \sin(\varphi + \theta)
$$
  
\n• Viscous forces  
\n
$$
\vec{F}_{v,n} = 6\pi \mu_{liq} R \frac{R}{H} \vec{v}_{rel,n}
$$
\n
$$
\vec{F}_{v,t} = 6\pi \mu_{liq} R (\frac{8}{15} \ln \frac{R}{H} + 0.9588) \vec{v}_{rel,t}
$$
  
\n• Critical rupture distance

**5**

#### **Wet Model Validation**

**NERGY** 

∃

**NATIONAL** ERG) TECHNOLOGY **ABORATORY** 



# **Effect of Liquid Volume**

**NATIONAL** ERG) **TECHNOLOGY ABORATORY** 



 $\mathbb{R}$ 

**7**

### **Effect of Liquid Viscosity**







#### **Effect of Liquid Surface Tension**





#### **U.S. DEPARTMENT OF IERGY** ∃

## **Liquid Bridge Model Implementation for Pyrolysis**





- The liquid bridge model in MFiX for cold flow was extended to implement a novel capability that explicitly models the mass, volume, and species of the liquid layer
- The evolution of the liquid bridge forces can be accurately modeled as the liquid volume changes (e.g., during pyrolysis)
- The "last species" volume is used to compute the capillary force instead of externally defined volume

volLL = DES X s(LL,NMAX(phaseLL)) \* PMASS(LL) / RO Xs0(phaseLL, NMAX(phaseLL))



#### **Transient Evolution of Liquid Layer Validation**



• Liquid layer in Tang et al.'s model is replaced with melted LDPE and allowed to pyrolyze to a gas pseudospecies: LDPE<sub>(liq)</sub> → Volatiles<sub>(gas)</sub>,  $A = 121.0 \cdot 10^9$  1/s,  $E = 159 \cdot 10^3$  J/mol



#### **Effect of Agglomeration on Pyrolysis**

ATIONAL TECHNOLOGY **ABORATORY** 

• Reaction mechanism from Ding et al.<sup>5</sup>/Encinar et al.<sup>6</sup> simplifies pyrolysis products to separate pseudospecies representing gas and oils: LDPE  $\rightarrow 0.4 \cdot$  Gas +  $0.6 \cdot$  Tar;  $0.0025$ Dotted lines: w/o LBM, Solid lines: w/ LBM  $A = 2.30 \cdot 10^{18}$  s<sup>-1</sup>;  $E = 285.7$  MJ/kmol; Δh = 975 kJ/kg **6** 0.002<br> **t d** 0.0015<br> **d** 0.001<br> **d** 0.0005 • Tar → Gas;  $A = 4.25 \cdot 10^6 \text{ s}^{-1}$ ;  $E = 108.0 \text{ MJ/kmol}$ ;  $\Delta h = -42 \text{ kJ/kg}$ 0.002 From left to right: Reacting w/o LBM, reacting w/ LBM Time: 0.05 Time: 0.05 s 0.0015 — Tar  $1.2e + 0.3$ Gas 1000 900 800 -0  $2e+02$ 10 20 30 40 50 60  $\sqrt{2}$ 1.5e-03 Time (s)  $0.001$ 0.0001 Dotted lines: w/o LBM, Solid lines: w/ LBM0.0005  $\sum_{8}^{3} 0.00008$  $0.0 + 00$ yield 0.00006 Temperature  $-$ Tar 800.0 850.0 900.0 973.15 723.15 Φ LDPE **RRate Pyrolysis** Cumulativ **Gas** le-10 le-9 le-8 le-7 le-6 le-50.0001 .0e-16 2e-12 4e-12 6e-12 8e-12 1.0e-11 0.00004 <u> 11110 - 1 1 11110 - 1 1 11110 - 1 1 11110 - 1 1 11110 - 1 1</u> 0.00002 ſ 10 20 30 40 50 60 0 Time (s) Time: 0.05 s

# **Continuous Operation Validation**

- Established CFD model of conical spouted bed reactor with continuous plastic feed based (loosely) on pilot-scale reactor of Aguado et al.<sup>7</sup>
	- As the plastic material is introduced into the reactor, it melts onto the sand particles and coats them
	- If the thickness of the layer that coats the particles is lower than a critical value, the sand particles do not fuse
	- Beyond this value, agglomerates grow irreversibly, and total blockage of the bed or defluidization is the result
- Good performance is determined on the basis of the critical thickness of the melted plastic that can be handled
- Sensitivity to solids holdup, material size/type, and fluidizing velocity can be compared





**ATIONAL** 

#### **Baseline Results with No Melting**

• Pyrolysis reaction occurs directly from plastic particles (no mass transfer to sand bed)





**ATIONAL** 

**NOLOGY** ORATORY

# **Results with Melting and Agglomeration**



• Plastic particles ≳200 °C are considered to be melted and allowed to fuse with the sand particles on contact; the liquid volume is subsequently redistributed between sand particles during sand-sand collisions



• Agglomeration due to cohesive liquid bridge forces causes the reactor to defluidize; the blockage dissipates when inlet velocity is increased from 3 m/s to 3.25 m/s after 2.5 s



# **Results with Melting and Agglomeration**

- The agglomeration-induced defluidization occurs during initial startup when particle temperatures and hence pyrolysis rates are low
- If the startup velocity is 3.25 m/s for 2.5 s and subsequently reduced to 3 m/s, the bed is hot enough such that continuous operation prevails even at the lower velocity



• As such, the MFiX model with liquid bridge implementation can help to optimize the operating envelop for minimal impact on the performance of the pyrolysis reactor



#### **Non-Instantaneous Liquid Distribution**



- Dr. Tafti simulated the actual collision/coating process between liquid plastic and sand for a range of important non-dimensional ratios of inertial, viscous, and capillary forces
- For liquid viscosity of 1 kg/m-s (1000 mPa-s), the liquid transfer time was around 1 ms
- To incorporate a finite (i.e., non-instantaneous) liquid transfer time during the redistribution of the liquid layer during particle separation, the amount of mass transferred is multiplied by a coefficient

```
xferMass = 1.0d0 * abs(massLL - massI) / 2.d0
```
• The coefficient could be determined as a ratio of the collision time to the liquid transfer time constant from Dr. Tafti



#### **Non-Instantaneous Liquid Distribution**

• Sensitivity to the mass transfer coefficient is determined for values of 1.0, 0.01, and 0.0001 (without reactions)



• The liquid transfer time is expected to be higher for more viscous plastics



### **Instantaneous Liquid Distribution + Pyrolysis**

- Repeat of earlier result with increased solids holdup and fixed mesh at different inlet temperatures compared with inert case
- Recall simplified reaction:  $\mathsf{LDFE}_{\mathsf{(liq)}} \to \mathsf{Volatiles}_{\mathsf{(gas)}}$  ,  $A = 121.0 \cdot 10^9$  1/s ,  $E = 159 \cdot 10^3$  J/mol





#### **Non-Instantaneous Liquid Distribution + Pyrolysis**

**ATIONAL HNOLOGY** 

• Mass transfer ratio = 0.01





#### **Non-Instantaneous Liquid Distribution + Pyrolysis**

**ATIONAL NOLOGY** 

• Mass transfer ratio = 0.0001





#### **References**



- 1. Grohn, P., Lawall, M., Oesau, T., Heinrich, S. and Antonyuk, S. (2020) CFD-DEM simulation of a coating process in a fluidized bed rotor granulator. *Processes*, 8(9): 1090.
- 2. Israelachvili, J.N. (2011) Intermolecular and Surface Forces; Elsevier: Amsterdam, The Netherlands.
- 3. Tang, T., He, Y., Ren, A., and Wang, T. (2019). Experimental study and DEM numerical simulation of dry/wet particle flow behaviors in a spouted bed. *Ind. Eng. Chem. Res.*, 58, 15353–15367.
- 4. Beetstra, R., van der Hoef, M.A., and Kuipers, J.A.M. (2007). Drag force of intermediate Reynolds number flow past mono- and bidisperse arrays of spheres. *AIChE J.*, 53, 489–501.
- 5. Ding, K. et al. (2020). CFD simulation of combustible solid waste pyrolysis in a fluidized bed reactor. *Powder Technol.*, 362, 177–187.
- 6. Encinar, J.M. and González, J.F. (2008). Pyrolysis of synthetic polymers and plastic wastes. Kinetic study. *Fuel Process. Technol.*, 89(7), 678–686.
- 7. Aguado, R., et al. (2005). Defluidization modelling of pyrolysis of plastics in a conical spouted bed reactor. *Chem. Eng. Process.*, 44, 231–235.



# **NETL RESOURCES**

VISIT US AT: **www.NETL.DOE.gov**

**@NETL\_DOE**

**@NETL\_DOE**



**@NationalEnergyTechnologyLaboratory**

CONTACT: Subhodeep Banerjee [Subhodeep.Banerjee@netl.doe.gov](mailto:Subhodeep.Banerjee@netl.doe.gov)

